

**Hydrochloric Acid**

**SECTION 1 – CHEMICAL PRODUCT AND COMPANY IDENTIFICATION**

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<b>Manufacturer's name and address:</b> Olin Corporation – Chlor Alkali Products Division  CLEVELAND, TN OFFICE 490 Stuart Road NE Cleveland, TN 37312-4918 U.S. • (423) 336-4850	<b>Supplier's name and address:</b> Olin Canada, ULC d/b/a Olin Chlor Alkali Products  MONTREAL, QC OFFICE 2020 University, Suite 2190 Montreal, Quebec H3A 2A5 Canada • (514) 397-6100
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<b>Product Name:</b>	Hydrochloric Acid		
<b>CAS#:</b>	7647-01-0	<b>Preparation date (M/D/Y):</b>	05/11/2010
<b>MSDS Code:</b>	HCl-e	<b>Revision date (M/D/Y):</b>	02/08/11
<b>Synonyms:</b>	Muriatic acid, Aqueous hydrogen chloride		
<b>Product Use:</b>	pH adjustment for water treatment, metal processing, sugar refining		

**Emergency Contacts (24 hr.)**

FOR INFORMATION REGARDING ON SITE CHEMICAL EMERGENCIES INVOLVING A SPILL OR LEAK, CALL

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**Canada: 1-800-567-7455**  
**U.S.: 1-800-424-9300 – CHEMTREC**

**SECTION 2 – COMPOSITION / INFORMATION ON INGREDIENTS**

Hazardous Ingredient(s)	% (w/w)	ACGIH Ceiling (TLV-C) / OSHA / (PEL-C)	CAS NO.
Hydrogen chloride	20 – 40	ACGIH 2 ppm OSHA 5 ppm	7647-01-0

**SECTION 3 – HAZARD IDENTIFICATION**

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**Emergency Overview:** Danger! Extremely corrosive! Causes severe burns and eye damage. Harmful if inhaled. Harmful or fatal if swallowed. Highly reactive with alkaline materials. Not flammable, but reacts with most metals to form explosive/flammable hydrogen gas. Read the entire MSDS for a more thorough evaluation of the hazards.  
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**Potential Health Effects:**

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**Routes of exposure:** Inhalation, skin contact, eye contact and ingestion.

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**General:** Hydrochloric acid (HCl) is a very strong acid. Solutions can be extremely corrosive. The severity of effects depends on the concentration of the solution and the duration of contact. In general, HCl solutions and mists with a pH of 3 or less are a significant health concern.

**Inhalation:** Vapor or mist from concentrated solutions can cause severe nasal irritation, sore throat, choking, coughing and difficulty breathing (50-100 ppm). Prolonged exposures can cause burns and ulcers to the nose and throat. Severe exposures (e.g. 1000-2000 ppm), for even a few minutes, can cause a life-threatening accumulation of fluid in the lungs (pulmonary edema). Symptoms of pulmonary edema such as shortness of breath can be delayed for several hours after the exposure.

**Skin Contact:** Corrosive! Concentrated solutions may cause pain and deep and severe burns to the skin. Prolonged and repeated exposure to dilute solutions often causes irritation, redness, pain and drying and cracking of the skin.

**Eye Contact:** Immediate pain, severe burns and corneal damage, which may result in permanent blindness. Low concentrations of vapor or mist (10-35 ppm) can be immediately irritating, causing redness.

**Ingestion:** Causes severe irritation or corrosive burns to mouth, throat, esophagus and stomach. Symptoms may include difficulty in swallowing, intense thirst, nausea, vomiting, diarrhea and in severe cases, collapse and death.

**Existing Medical Conditions Possibly Aggravated by Exposure:** Skin irritation may be aggravated in individuals with existing skin lesions. Breathing of vapors or sprays (mists) may aggravate acute or chronic asthma and chronic pulmonary disease such as emphysema and bronchitis.

**Chronic Effects:** Repeated exposure to low concentrations of acid mist or vapor may cause redness, swelling and pain (dermatitis). Exposure to low concentration of acid mist or vapor by inhalation may cause bleeding of nose and gums, bronchitis, stomach pain (gastritis), and brownish discoloration and damage to tooth enamel. Dental erosion becomes more severe with increased exposure.

**Carcinogenicity:** Hydrochloric acid is not classified as carcinogenic by ACGIH (American Conference of Governmental Industrial Hygienists) or IARC (International Agency for Research on Cancer), not regulated as carcinogens by OSHA (Occupational Safety and Health Administration), and not listed as carcinogens by NTP (National Toxicology Program).

**IARC Evaluation:** There is inadequate evidence for the carcinogenicity in humans of hydrochloric acid. There is inadequate evidence for the carcinogenicity in experimental animals of hydrochloric acid. Overall evaluation: Hydrochloric acid is not classifiable as to its carcinogenicity to humans (Group 3).

△ **Other important hazards:** Refer to TOXICOLOGICAL INFORMATION (Section 11) for additional information.

### SECTION 4 – FIRST AID MEASURES

**General:** Corrosive effects on the skin and eyes may be delayed, and damage may occur without the sensation or onset of pain. Strict adherence to first aid measures following any exposure is essential. SPEED IS ESSENTIAL. OBTAIN IMMEDIATE MEDICAL ATTENTION.

**Inhalation:** Move victim to fresh air. Give artificial respiration ONLY if breathing has stopped. Do not use mouth-to-mouth method if victim ingested or inhaled the substance: induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Give Cardiopulmonary Resuscitation (CPR) if there is no pulse AND no breathing. Obtain medical attention IMMEDIATELY. Symptoms may appear up to 48 hrs after exposure.

**Skin Contact:** Immediately flush skin with running water for a minimum of 20 minutes. Start flushing while removing contaminated clothing. If irritation persists, repeat flushing. Obtain medical attention

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IMMEDIATELY. Do not transport victim unless the recommended flushing period is completed or flushing can be continued during transport.

While the patient is being transported to a medical facility, apply compresses of iced water. If medical treatment must be delayed, immerse the affected area in iced water. If immersion is not practical, compresses of iced water can be applied. Avoid freezing tissues.

Discard heavily contaminated clothing and shoes in a manner, which limits further exposure. Otherwise, wash clothing separately before reuse.

**Eye Contact:** Immediately flush eyes with running water for a minimum of 20 minutes. Hold eyelids open during flushing. If irritation persists, repeat flushing. Obtain medical attention IMMEDIATELY. Do not transport victim until the recommended flushing period is completed unless flushing can be continued during transport.

**Ingestion:** DO NOT INDUCE VOMITING. If victim is alert and not convulsing, rinse mouth and give 240 to 300 mL (8 to 10 oz.) of water to dilute material. If spontaneous vomiting occurs, have victim lean forward with head down to avoid breathing in of vomitus, rinse mouth and administer more water. IMMEDIATELY contact local poison control center. IMMEDIATELY transport victim to an emergency facility. Never give anything by mouth if victim is rapidly losing consciousness, or is unconscious or convulsing.

**SECTION 5 – FIRE FIGHTING MEASURES**

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<b>Flash Point and method</b>	Not applicable. Not combustible
<b>Flammable Limits (Lower)</b>	Not applicable
<b>Flammable Limits (Upper)</b>	Not applicable
<b>Auto Ignition Temperature</b>	Not applicable
<b>Decomposition Temperature</b>	Thermally stable up to temperatures of about 1500°C (2730°F).
<b>Combustion and Thermal Decomposition Products</b>	Hydrogen and chlorine
<b>Rate of Burning</b>	Not applicable
<b>Explosive Power</b>	Not sensitive
<b>Sensitivity to Mechanical Impact</b>	Not sensitive
<b>Sensitivity to Static Discharge</b>	Not sensitive
<b>Fire and Explosion Hazards</b>	Reacts with many metals to liberate hydrogen gas, which can form explosive mixtures with air. Hydrogen, a highly flammable gas, can accumulate to explosive concentrations inside drums, or any types of steel containers or tanks upon storage.
<b>Extinguishing media</b>	For large fires use extinguishing agents compatible with acid and appropriate for the burning material. An all purpose type AFFF foam may be used according to foam manufacturer's recommended techniques. The foam supplier should be consulted for recommendations regarding foam types and delivery rates for specific applications. Use carbon dioxide or dry chemical media for small fires. Do NOT use carbon dioxide, if cyanides are involved in fire. If only water is available, use it in the form of a fog.

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**Fire Fighting Procedures:** As appropriate for surrounding materials/equipment. Water spray should be used to cool containers. Water spray may be used to knock down escaping vapor.

**Fire Fighting Protective Equipment:** Use self-contained breathing apparatus and special protective clothing.

**Evacuation:** If tank, rail car or tank truck is involved in a fire, ISOLATE for 800 meters (1/2 mile) in all directions; also, consider initial evacuation for 800 meters (1/2 mile) in all directions.

**NOTE:** Also see "Section 10 - Stability and Reactivity"

**SECTION 6 – ACCIDENTAL RELEASE MEASURES**

**Spills, Leaks, or Releases:**

- Restrict access to area until completion of clean up. Ensure trained personnel conduct clean up.
- Wear adequate personal protective equipment. Do not touch spilled material.
- Remove all ignition sources (no smoking, flares, sparks or flames). All equipment should be grounded. Ventilate area.
- Stop leak if possible without personal risk.
- Small spills: Cover with DRY earth, sand or other non-combustible material. Use clean non-sparking tools to collect material and place it into loosely covered plastic containers for later disposal.
- Large spills: Isolate spill or leak area immediately for at least 50 meters (160 feet) in all directions. Keep unauthorized personnel away. Stay upwind. Keep out of low areas. Prevent entry into sewers and confined areas. Dike with inert material (sand, earth, foamed polyurethane, foamed concrete, etc.). Consider in-situ neutralization and disposal. Absorb bulk liquid with fly ash or cement powder. Neutralize with recommended materials, taking care to avoid any foaming or splattering that may occur from the neutralization reaction of the acid with these materials. Make sure all liquid has been thoroughly contacted and absorbed by the dry materials. Transfer absorbed spill material and any contaminated underlying soil to a suitable chemical waste container. Ensure adequate decontamination of tools and equipment following clean up. Washing down of spills with water is not recommended as this tends to spread the contamination and increases the likelihood of percolating the acid down through the soil and/or of uncontrolled flow of acid into sewers, streams, or other waters. Hydrochloric acid leaks, or spills must not come in contact with any acid soluble sulfide wastes (such as sewers) because of the danger of evolving hydrogen sulfide gas.

Comply with Federal, Provincial/State and local regulations on reporting releases.

**Deactivating Chemicals:** Lime, limestone, sodium carbonate (soda ash), sodium bicarbonate. The following absorbent materials have been tested and recommended for vapor suppression and/or containment of 26% and 35% hydrochloric acid solutions: a mixture of (75%) anionic polyacrylamide (R1779) and (25%) nonionic polyacrylamide (Versicol W25), individually use the anionic polyacrylamide or nonionic polyacrylamide, and Cellosize WP3H (hydroxyethyl cellulose).

**Waste Disposal Methods:** Dispose of waste material at an approved waste treatment/disposal facility, in accordance with applicable regulations. Do not dispose of waste with normal garbage or to sewer systems.

- Note:**
- Clean-up material may be a RCRA Hazardous Waste on disposal.
  - Spills are subject to CERCLA reporting requirements: RQ = 5000 lbs. (≈ 500 gal.; 2270 kg).

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## SECTION 7 – HANDLING AND STORAGE

**Handling:** Take all precautions to avoid personal contact. Prevent release of vapor or mist into workplace air. Always ensure adequate ventilation in handling areas. Locate safety shower and eyewash station close to chemical handling area. Inspect containers for leaks before handling. Use EXTREME care when diluting with water. Always add acid to water. CAUTION: Hydrogen, a highly flammable gas, can accumulate to explosive concentrations inside drums, or any types of steel containers or tanks upon storage. Storage containers should be vented on a regular basis by trained personnel ONLY. Label containers. Keep containers closed when not in use. Empty containers may contain residues, which are hazardous.

**Storage:** Store in a cool, dry, well ventilated area, out of direct sunlight and away from heat sources. Store away from incompatible materials such as oxidizing materials, reducing materials and strong bases. Use corrosion-resistant structural materials and lighting and ventilation systems in the storage area. Use containers, which are securely labeled and protected from damage. Storage drums must be coated with an acid resistant material. Rubber-lined steel, PVC/FRP, FRP, Hastelloy C-276, Inconel 625, and tantalum, are the most commonly used corrosion - resistant materials of construction at room temperature. Rubber, glass, plastic and ceramic ware are also resistant to corrosion. Vented containers must be used and must be kept closed when not being used. Containers should have a safety relief valve. Care should be taken to release any internal pressure slowly. Use corrosion-resistant transfer equipment when dispensing. Limit quantity of material in storage. Restrict access to storage area. Post warning signs when appropriate. Keep storage area separate from populated work areas. Inspect periodically for deficiencies such as damage or leaks.

Storage tanks should be above ground and surrounded with dikes capable of holding entire contents.

**Storage Temperature:** Exposure to extremes of heat and cold should be avoided. Ideal storage temperature is 10-27°C (50-80.6°F). Do not expose sealed containers to temperatures above 40°C (104°F).

**Other Precautions:** If stored indoors, building floors should be acid resistant with drains to a recovery tank. Electrical equipment should be flameproof and protected against corrosive action. Wood and other organic materials should not be used on floors, structural materials and ventilation systems in the storage area.

## SECTION 8 – EXPOSURE CONTROLS / PERSONAL PROTECTION

### PREVENTIVE MEASURES

Recommendations listed in this section indicate the type of equipment, which will provide protection against over exposure to this product. Conditions of use, adequacy of engineering or other control measures, and actual exposures will dictate the need for specific protective devices at your workplace.

**Engineering Controls:** Local exhaust ventilation should be applied wherever there is an incidence of point source emissions or dispersion of regulated contaminants in the work area. The most effective measures are the total enclosure of processes and the mechanization of handling procedures to prevent all personal contact with hydrochloric acid. Because of the high potential hazard associated with this substance, stringent control measures such as enclosure or isolation are recommended when dealing with large quantities. Electrical installations should be protected against the corrosive action of acid vapors. Smoking should be prohibited in areas in which hydrochloric acid is stored or handled.

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### **PERSONAL PROTECTIVE EQUIPMENT**

**Eye Protection:** Wear splash resistant chemical goggles and full faceshield. Maintain eye wash fountain and quick-drench facilities in work area.

**Skin Protection:** Wear impervious protective clothing, including boots, gloves, lab coat, apron or full body suit, as appropriate, to prevent skin contact.

**Recommended Materials:** Guidelines for hydrochloric acid, 37%:

- RECOMMENDED (resistance to breakthrough longer than 8 hours): Butyl rubber, neoprene, Viton™, Saranex™, Barricade, CPF 3™, Responder™, Trellechem HPS™.
- RECOMMENDED (resistance to breakthrough longer than 4 hours): Natural rubber, nitrile rubber, polyvinyl chloride, Teflon™, 4H™ (polyethylene/ethylene vinyl alcohol).
- CAUTION, use for short periods only (resistance to breakthrough less than 1 hour): Polyethylene.
- NOT RECOMMENDED for use (resistance to breakthrough less than 1 hour): Polyvinyl alcohol.

### **Respiratory Protection:**

Up to 50 ppm: Chemical cartridge respirator with cartridge(s) to protect against hydrogen chloride; or gas mask with canister to protect against hydrogen chloride or powered air-purifying respirator with cartridge(s) to protect against hydrogen chloride; or Supplied Air Respirator (SAR); or full-facepiece Self-contained breathing apparatus (SCBA).

EMERGENCY or planned entry into unknown concentration or IDLH conditions: Positive pressure, full-facepiece SCBA; or positive pressure, full-facepiece SAR with an auxiliary positive pressure SCBA.

ESCAPE: Gas mask with acid gas canister or escape-type SCBA.

### **EXPOSURE GUIDELINES**

#### **PRODUCT: Hydrochloric Acid:**

ACGIH Ceiling Exposure Limit (TLV-C) 2 ppm (3 mg/m<sup>3</sup>)  
OSHA Ceiling Exposure Limit (PEL-C): 5 ppm (7 mg/m<sup>3</sup>)  
NIOSH IDLH 50 ppm

AIHA- Emergency Response Planning Guidelines (ERPGs)

*ERPGs are for community emergency planning limits and not workplace exposure limits.*

ERPG-1: 3 ppm

ERPG-2: 20 ppm

ERPG-3: 150 ppm

The **ERPG-1** is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing other than mild transient adverse health effects or perceiving a clearly defined, objectionable odor.

The **ERPG-2** is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing or developing irreversible or other serious health effects or symptoms, which could impair an individual's ability to take protective action.

The **ERPG-3** is the maximum airborne concentration below which it is believed that nearly all individuals could be exposed for up to 1 hr without experiencing or developing life-threatening health effects.

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**SECTION 9 – PHYSICAL AND CHEMICAL PROPERTIES**

<b>Alternate Name(s)</b>	Hydrogen chloride, Muriatic acid
<b>Chemical Name</b>	Hydrochloric acid
<b>Chemical Family</b>	Inorganic acid
<b>Molecular Formula</b>	H-Cl
<b>Molecular Weight</b>	36.46
<b>Physical State and Appearance</b>	Colorless, or slightly yellow liquid
<b>Odor</b>	Pungent odor
<b>pH</b>	Less than 1
<b>Solubility (Water)</b>	Miscible in all proportions in water
<b>Solubility (Other)</b>	Soluble in alcohol, ethers , benzene.
<b>% Volatile by Volume</b>	100
<b>% Volatile Organic Compounds</b>	Zero
<b>Coefficient of OIL/Water Distribution</b>	Not Available

BAUME	13°Be	18°Be	20°Be	22°Be	23°Be
<b>Concentration</b>	19.63	27.92	31.45	35.21	37.14
<b>Vapor Pressure (mm Hg at 20°C)</b>	0.3	11	20	72	150
<b>Boiling Point (°C):</b>	109	98	85	62	50
<b>Melting Point (°C):</b>	-55	-58	-40	-31	-27
<b>Freezing Point (°C):</b>	-55	-58	-40	-31	-27
<b>Specific Gravity</b>	1.10	1.14	1.16	1.18	1.19
<b>Viscosity (cp at 20°C):</b>	1.28	1.60	1.75	1.90	2.00

**SECTION 10 – STABILITY AND REACTIVITY**

**Hazardous Decomposition Products:** When heated to decomposition, emits toxic hydrogen chloride fumes. Thermal oxidative decomposition produces toxic chlorine fumes and explosive hydrogen gas.

**Chemical Stability:** Stable under conditions of normal use.

**Conditions to Avoid:** Avoid heat, flames, sparks and other sources of ignition.

**Incompatibility with other Substances:** A strong mineral acid, concentrated hydrochloric acid is incompatible with many substances and highly reactive with strong bases, metals, metaloxides, hydroxides, amines, carbonates and other alkaline materials. Incompatible with materials such as cyanides, sulfides, sulfites, sulfuric acid and formaldehyde. Contact with metals may produce flammable hydrogen gas. When diluting, add acid to water. Do NOT add water to the acid.

**Hazardous Polymerization:** Will not occur. Hydrochloric acid is a stable product and does not polymerize. However, it may induce hazardous polymerization with aldehydes and epoxides.

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## SECTION 11 – TOXICOLOGICAL INFORMATION

Δ For additional toxicological information, refer to Section 3.

### **TOXICOLOGICAL DATA:**

Δ **Toxicological Data:**

LD<sub>50</sub> (oral, rat) = 700 mg/kg

LD<sub>50</sub> (dermal, rabbit) = >5010 mg/kg

LC<sub>50</sub> (inhalation, rat) = 1562 ppm for 4 hr (3124 ppm for 1 hr)

LC<sub>50</sub> (inhalation, mouse) = 1108 ppm/1 hr

Eye Effects (rabbit): Application of a 1% hydrochloric acid (0.25N) solution for 20 seconds caused scarring of the cornea. Other studies have reported that application of 5 mg for 30 seconds caused mild irritation, and that application of a 5% solution caused minimal irritation (duration not indicated).

Skin Effects (rabbit): Application of 0.5 mL of a 17% concentrated solution for 4 hours caused corrosive burns.

Δ **Sensitization to material:** Not expected to cause respiratory or skin sensitization reactions.

**Mutagenicity:** Mutagenic effects have been reported in one bacterial test (E. Coli-DNA repair), in three insect tests (Drosophila, grasshopper) and in one in vitro mammalian cell test (hamster lung cells). HCl was negative in another in vitro mammalian cell test (Syrian Hamster Embryo cells). The significance of the positive reports is questionable since pH (acidity) can influence the results of short-term tests.

**Reproductive Effects:** Female rats were exposed to 450 mg/m<sup>3</sup> for 1 hour either prior to mating or on day 9 of pregnancy. Developmental effects were observed in the offspring. However, this exposure caused toxic effects, including mortality, in the mothers.

**Teratogenicity and Fetotoxicity:** No information is available.

**Synergistic Materials:** None known

## SECTION 12 – ECOLOGICAL INFORMATION

### **Ecotoxicological Information:**

Fish Toxicity: LC<sub>100</sub> Trout 10 mg/L 24hr,  
LC<sub>50</sub> Shrimp 100 to 330 ppm/48 hr (salt water)  
LC<sub>50</sub> Gold fish 178 mg/L (one to two hours of survival time),  
TLm/mosquito fish/ 282 ppm/96 hr/fresh water

The concentration of hydrochloric acid that was found to be injurious to crops is 350 mg/L.

Toxicity is primarily associated with pH. Toxic to aquatic life.

Invertebrate and Microbial Toxicity: Acidification of soy broth containing *Listeria monocytogenes* to pH 4.4 inhibited microbial activity.

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**Persistence and Degradation:** When hydrochloric acid is spilled onto soil, it will begin to infiltrate. The presence of water in the soil will influence the rate of chemical movement in the soil. During transport through the soil, hydrochloric acid will dissolve some of the soil material, in particular those of a carbonate base. The acid will be neutralized to some degree. However, significant amounts of acid are expected to remain for transport down toward the ground water table. Hydrogen chloride in water dissociates almost completely, with the hydrogen ion captured by the water molecules to form the hydronium ion.

**SECTION 13 – DISPOSAL CONSIDERATIONS**

Review federal, state and local government requirements prior to disposal.

Do not dispose of waste with normal garbage, or to sewer systems.

Whatever cannot be saved for recovery or recycling, including containers, should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options.

**RCRA:** Test waste material for corrosivity, D002, prior to disposal.

**SECTION 14 – TRANSPORT INFORMATION**

	TDG CLR *	DOT
Shipping Name	Hydrochloric Acid	Hydrochloric acid
Hazard Class / Division	8	8
Identification No. Packing Group:	UN1789 II	UN1789 II
ERAP/ RQ	3000 L	RQ = 5000 lbs. (2270 Kg)

**Note: \* TDG CLR (Clear Language Regulations) became effective August 15, 2002**

TDG – Emergency Response Assistance Plan (ERAP requirements of part 7 must be met for quantities exceeding 3000 liters per consignment.

- Δ **IATA/ICAO Shipping Description:** UN1789, Hydrochloric acid, Class 8, PG II is accepted for air transport.
- Δ **For Chemical Emergencies in Transportation Requiring Activation of Olin 24 Hour Emergency Response Plan Call:**
  - U.S. 1-800-424-9300 – Chemtrec
  - Canada 1-800-567-7455

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**SECTION 15 – REGULATORY INFORMATION**

**Δ CANADIAN INFORMATION:**

This product has been classified in accordance with the hazard criteria of the CPR (Controlled Products Regulations) and this MSDS (Material Safety Data Sheet) contains all the information required by the CPR.

**Controlled Products Regulations (WHMIS) Classification:**

Class D1A – Immediate and serious effects – Very Toxic  
Class E – Corrosive

**CEPA / Canadian Domestic Substances List (DSL):** On the Canadian Domestic Substances List (CEPA DSL).

**WHMIS Ingredient Disclosure List:** Meets criteria for disclosure at 1% or greater.

National Pollutant Release Inventory (NPRI): Y

**Δ USA INFORMATION:**

**OSHA Classification:** Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200)

**SARA Regulations sections 313 and 40 CFR 372:** Y

**SARA Hazard Categories, SARA SECTIONS 311/312 (40 CFR 370.21):**

ACUTE: Y  
CHRONIC: N  
FIRE: N  
REACTIVE: N  
SUDDEN RELEASE: Y  
OSHA PROCESS SAFETY (29 CFR 1910.119): Y

**CERCLA 40 CFR 302.4:** Y

Reportable Quantity = 5000 lbs (2270 kg)

**TSCA Inventory Status:** Y

This product does not contain nor is it manufactured with ozone depleting substances.

**Other Regulations/Legislation which apply to this product:**

California Director's List of Hazardous Substances, Rhode Island Hazardous Substance List, New Jersey Environmental Hazardous Substance, Minnesota Hazardous Substance List, Massachusetts Extraordinarily Hazardous Substance, Florida Hazardous Substances List.

**Right –To-Know:** Illinois, Massachusetts, New Jersey, Pennsylvania

**Δ EUROPEAN ECONOMIC COMMUNITY (EEC) INFORMATION:**

**EEC Classification:** C, R 34 - 37

**EINECS:** 231-595-7

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**CALIFORNIA PROP 65 COMPONENTS:**

This product is not listed, but it may contain elements known to the State of California to cause cancer or reproductive toxicity as listed under Proposition 65 State Drinking Water and Toxic Enforcement Act. For additional information, contact Olin Technical Services (800-299-6546)

**SECTION 16 – OTHER INFORMATION**

- Δ The information contained herein is offered only as a guide to the handling of this specific material and has been prepared in good faith by technically knowledgeable personnel. It is not intended to be all-inclusive and the manner and conditions of use and handling may involve other and additional considerations. No warranty of any kind is given or implied and Olin will not be liable for any damages, losses, injuries or consequential damages that may result from the use of or reliance on any information contained herein. This Material Safety Data Sheet is valid for three years.

**Revision Indicators:**

- Δ In the left margin indicates a revision or addition of information since the previous issue.

**National Fire Protection Association (NFPA) Rating  
Hazardous Materials Identification System (HMIS) Rating**

	NFPA	HMIS	
HEALTH	3	3	4 = Extreme/Severe
FIRE	0	0	3 = High/Serious
REACTIVITY / INSTABILITY	1	1	2 = Moderate
SPECIAL HAZARDS	N/Ap	N/Ap	1 = Slight
			0 = Minimum
			W = Water Reactive
			OX = Oxidizer
			* = Chronic health hazard

Δ **REFERENCES:**

1. RTECS-Registry of Toxic Effects of Chemical Substances, Canadian Centre for Occupational Health and Safety RTECS database, National Institute for Occupational Safety and Health, U.S. Dept. of Health and Human Services, Cincinnati, 2008
2. Transport Of Hazardous Materials (49CFR), Canadian Centre for Occupational Health and Safety, (2008)
3. "CHEMINFO", Canadian Centre for Occupational Health and Safety, Hamilton, Ontario, Canada, (2008).
4. Chemlist, STN Database, Chemical Abstract Service, (2005)
5. Chemical Hazards Response Information System (CHRIS), CCOHS, (2008).
6. HSDB-Hazardous Substances Data Bank, through "CCINFO disc", Canadian Centre for Occupational Health and Safety, Hamilton, Ontario, Canada, (2008).
7. NFPA 49 Hazardous Chemicals Data 1994 Edition, National Fire Protection Association, Quincy, MA, 1994.
8. NIOSH POCKET GUIDE TO CHEMICAL HAZARDS, U.S. Department of Health and Human Services, National Institute for Occupational Safety and Health, June 1997.
9. "2008 Threshold Limit Values and Biological Exposure Indices", American Conference of Government Industrial Hygienists, 2008.
10. TRANSPORT OF DANGEROUS GOODS (TDG), Canadian Centre for Occupational Health and Safety, (2008)

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Δ **LEGEND**

- ACGIH - American Conference of Governmental Industrial Hygienists
- AFFF - Aqueous Film Forming Foam
- AIHA - American Industrial Hygiene Association
- CAS # - Chemical Abstracts Service Registry Number
- CERCLA - Comprehensive Environmental Response, Compensation, and Liability Act
- CFR - Code of Federal Regulations
- DOT - Department of Transportation
- EINECS - European Inventory of Existing Chemical Substances
- EPA - Environmental Protection Agency
- ERAP - Emergency Response Assistance Plan
- IATA - International Air Transportation Association
- ICAO - International Civil Aviation Organization
- FRP - Fiberglass Reinforced Plastic
- HMIS - Hazardous Materials Identification System
- IARC - International Agency for Research on Cancer
- IDLH - Immediately Dangerous to Life and Health
- LC<sub>50</sub> - The concentration of material in air expected to kill 50% of a group of test animals
- LD<sub>50</sub> - Lethal Dose expected to kill 50% of a group of test animals
- MSHA - Mine Safety and Health Administration
- N/Ap - Not Applicable
- N/Av - Not Available
- NFPA - National Fire Protection Association
- NIOSH - National Institute for Occupational Safety and Health
- NTP - National Toxicology Program
- OSHA - Occupational Safety & Health Administration
- PEL - Permissible Exposure Limit
- PVC - Polyvinyl chloride
- RCRA - Resource Conservation and Recovery Act
- SARA - Superfund Amendments and Reauthorization Act of the U.S. EPA
- STEL - Short Term Exposure Limit
- TDG - Transportation of Dangerous Goods Act/Regulations
- TLV - Threshold Limit Value
- TSCA - Toxic Substances Control Act
- TWA - Time Weighted Average
- WEEL - Workplace Environmental Exposure Level
- WHMIS - Workplace Hazardous Materials Identification System